

Dutch  
Metrology  
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WP2 First time isotope ratio gas reference materials for  $\delta^{13}\text{C}\text{H}_4$  and  $\delta^2\text{H}\text{-CH}_4$

Status report & upcoming activities

Stefan Persijn (VSL) on behalf of all WP2 partners

*Stakeholder catchup Stellar, 26 February 2021*

NPL, MPG, RUG, AL, JSI, Empa, UEF, INRIM, TUBITAK

# WP2 First time isotope ratio gas reference materials for $\delta^{13}\text{C-CH}_4$ and $\delta^2\text{H-CH}_4$

## Aim

Provide  $\text{CH}_4$  reference materials linked to the

## Current status

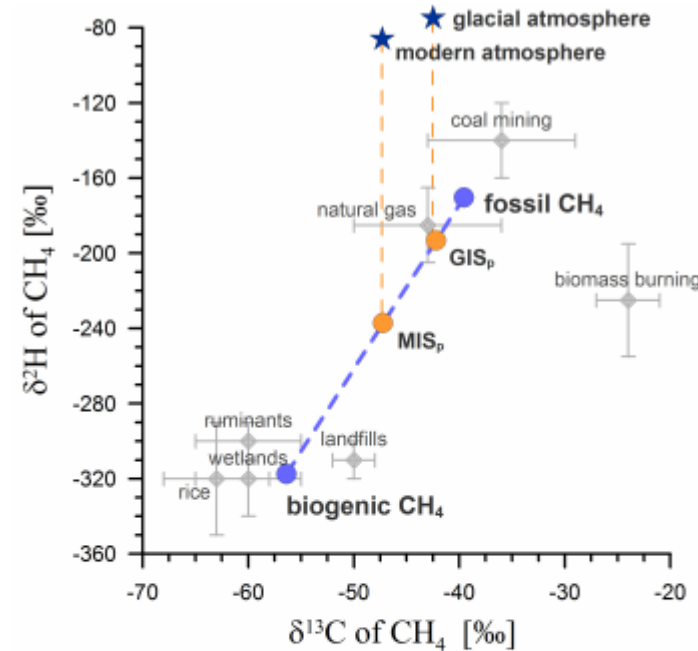
**Interlaboratory comparison of  $\delta^{13}\text{C}$  and  $\delta\text{D}$  measurements of atmospheric  $\text{CH}_4$  for combined use of data sets from different laboratories**

Taku Umezawa<sup>1,2</sup>, Carl A. M. Brenninkmeijer<sup>1</sup>, Thomas Röckmann<sup>3</sup>, Carina van der Veer<sup>4</sup>, Ryo Eniita<sup>6</sup>, Shinji Morimoto<sup>6,7</sup>, Shunji Aoki<sup>6</sup>, Todd Sowere<sup>8</sup>, Loeben Schmitt<sup>9</sup>, Michael Ro

<https://doi.org/10.5194/amt-11-1207-2018>

## Project will provide

To develop gas reference materials of methane (pur) with a repeatability of 0.02 ‰ for  $\delta^{13}\text{C-CH}_4$  and 1 ‰ for  $\delta^2\text{H-CH}_4$ . The target is to achieve a  $\delta^{13}\text{C-CH}_4$  and 5 ‰ for  $\delta^2\text{H-CH}_4$ , ensuring traceability to more than one year.



**Fig. 1.** Dual isotope signatures of  $\text{CH}_4$  for  $\delta^{13}\text{C}$  and  $\delta^2\text{H}$ . Grey diamonds mark the field of  $\delta^{13}\text{C}$  and  $\delta^2\text{H}$  isotopes of  $\text{CH}_4$  according to its source (Quay et al., 1999). Blue circles indicate  $\delta^{13}\text{C}$  and  $\delta^2\text{H}$  pairs of the parental  $\text{CH}_4$  gases (fossil and biogenic  $\text{CH}_4$ ). Based on our fossil and biogenic  $\text{CH}_4$ , we can produce filial  $\text{CH}_4$  mixtures with  $\delta^{13}\text{C}$  and  $\delta^2\text{H}$  isotope values that fall on the dashed blue mixing line. The two filial  $\text{CH}_4$  gas mixtures are indicated by orange circles where  $\text{GIS}_p$  and  $\text{MIS}_p$  represent the  $\delta^{13}\text{C}$  of glacial and modern atmospheric samples, respectively. Isotope signatures of glacial and modern atmospheric  $\text{CH}_4$  are indicated by the dark blue stars.

Source: Sperlich et a., 2012

o scales.

t offsets of  $\delta^{13}\text{C-CH}_4$  and  $\delta\text{D-CH}_4$  data sets reported from different laboratories at modern read over ranges of 0.5 ‰ for  $\text{CH}_4$ . The intercomparison re-



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# Task 2.1: Inventory of source and supply and development of pure methane gas reference materials

Aim: to identify commercially available pure CH<sub>4</sub> relevant for underpinning atmospheric measurements for source apportionment. Pure CH<sub>4</sub> will be used to prepare gas reference materials of pure CH<sub>4</sub>.

Isotope	Range suggested	Target repeatability	Target uncertainty
δ <sup>13</sup> C-CH <sub>4</sub>	-20 ‰ to -80 ‰	0.02 ‰	0.2 ‰
δ <sup>2</sup> H-CH <sub>4</sub>	-75‰ to -400 ‰	1 ‰	5 ‰

Key challenges:

- find proper cylinder passivation to enable long term stability and minimise fractionation
- a full understanding of the major contributors to the uncertainty budget
- high precision IRMS measurements to achieve traceability to the scales with low uncertainty

VSL, NPL, AL, MPG, JSI, RUG  
Start: Sep 2020

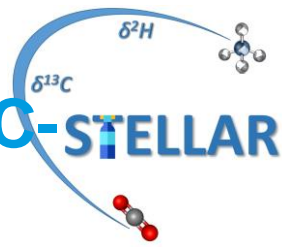
# Task 2.2: Diluted $\delta^{13}\text{C}\text{CH}_4$ and $\delta^2\text{H}\text{-CH}_4$ in air gas reference materials

Aims:

- 1) to prepare  $\geq 3$  reference gas mixtures of  $\text{CH}_4$  in a  $\text{CH}_4$ -free air matrix at  $1.85 \mu\text{mol mol}^{-1}$  with different isotopic composition, of which the isotope ratios and amount fractions have assigned values. Prepared by gravimetry in high pressure cylinders with same challenging target uncertainty as the pure  $\text{CH}_4$  and  $\geq 1$  year stability
  
- 2) fully characterize the zero-air used in the preparation to study the influence on fractionation of the purification process used to remove methane.
  
- 3) to prepare diluted  $\text{CH}_4$  reference materials in different matrix gases and at different amount of substance fractions to support research in WP3 towards novel calibration approaches and to study the influence of matrix gas changes on the uncertainty using spectroscopy.

VSL, NPL, AL, Empa, MPG, JSI, UEF, RUG  
 Start: June 2021

# Task 2.3: Linking to the $\delta^{13}\text{C}$ - $\text{CO}_2$ reference materials for an independent assessment of the accuracy and uncertainty of $\delta^{13}\text{C}$ -STELLAR $\text{CH}_4$ reference materials



Aim: to compare the  $\delta^{13}\text{C}$ - $\text{CH}_4$  reference materials to the  $\delta^{13}\text{C}$ - $\text{CO}_2$  reference materials to provide an independent assessment of the accuracy and uncertainty claims. A facility will be developed for the conversion of  $\text{CH}_4$  to  $\text{CO}_2$  and the sensitivities and contribution to uncertainty of this process will be fully characterised. The target uncertainty is 0.1 ‰ for  $\delta^{13}\text{C}$ - $\text{CO}_2$ .

NPL, INRIM, TUBITAK, RUG, MPG  
Start: Sep 2020

Month	1	2	3	4	5	6	7	8	9	10	11	12	13	14	15	16	17	18	19	20	21	22	23	24	25	26	27	28	29	30	31	32	33	34	35	36
Activities	Sep-20	Oct-20	Nov-20	Dec-20	Jan-21	Feb-21	Mar-21	Apr-21	May-21	Jun-21	Jul-21	Aug-21	Sep-21	Oct-21	Nov-21	Dec-21	Jan-22	Feb-22	Mar-22	Apr-22	May-22	Jun-22	Jul-22	Aug-22	Sep-22	Oct-22	Nov-22	Dec-22	Jan-23	Feb-23	Mar-23	Apr-23	May-23	Jun-23	Jul-23	Aug-23
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**Identify and obtain pure CH<sub>4</sub>**

**Linking to the δ<sup>13</sup>C-CO<sub>2</sub> reference materials**

Pure CH<sub>4</sub> gas standards

CH<sub>4</sub> gas standards at ambient levels

Link WP1-WP2

# Composition measurements of biogas

Main molecular composition (in mol%) of investigated biogas samples and natural gas samples

Determined  $\delta^{13}\text{C}$ -IRMS values of biogas  $\text{CH}_4$  fractions and natural gas samples after combustion to  $\text{CO}_2$ .

<i>Biogas</i>	$\text{CH}_4$ (mol%)	$\text{CO}_2$ (mol%)	$\text{C}_x\text{H}_y$ (mol%)	$(\text{N}_2, \text{O}_2, \dots)$ (mol%)	$\delta^{13}\text{C}$ (‰)
Maize_onions	56,3	42,2	0,02	1,5	-55,6
Landfill_1	61,1	32,1	0	6,8	-61,2
Organic waste	61,5	38,4	0	0,2	-52,0
Cookies_fish	85,2	12,0	0	2,8	-53,6
Mun. sewage sludge	89,5	10,3	0	0,3	-28,6
Landfill_2	54,4	30,5	0	15,1	-56,3
Sugar beet	87,8	8,2	0	4,1	-39,2
Manure_vegetables	59,6	35,8	0	4,7	-48,4
Manure_meat waste	96,4	3,3	0	0,4	-57,2
<i>Natural gas</i>					
Norway gas	87,1	2,06	8,9	1,95	-39,9
North Sea gas	85,07	2,66	4,93	7,34	-29,8
Groningen gas	81,05	1,02	3,45	14,48	-28,4

*sources:*

Palstra, S. W. L. and Meijer, H. A. J.: Biogenic Carbon Fraction of Biogas and Natural Gas Fuel Mixtures Determined 2014. with C-14, Radiocarbon, 56(1), 7–28, doi:10.2458/56.16514,

Palstra, S.W.L., et al. EMPIR project '16ENG05 Metrology for Biomethane' WP3.7 final report

# VSL activities to obtain pure CH<sub>4</sub>

## Fossil

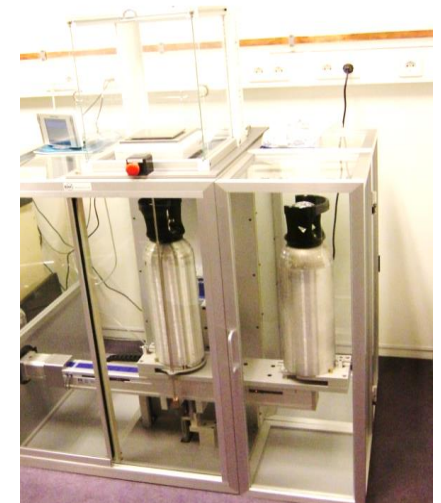
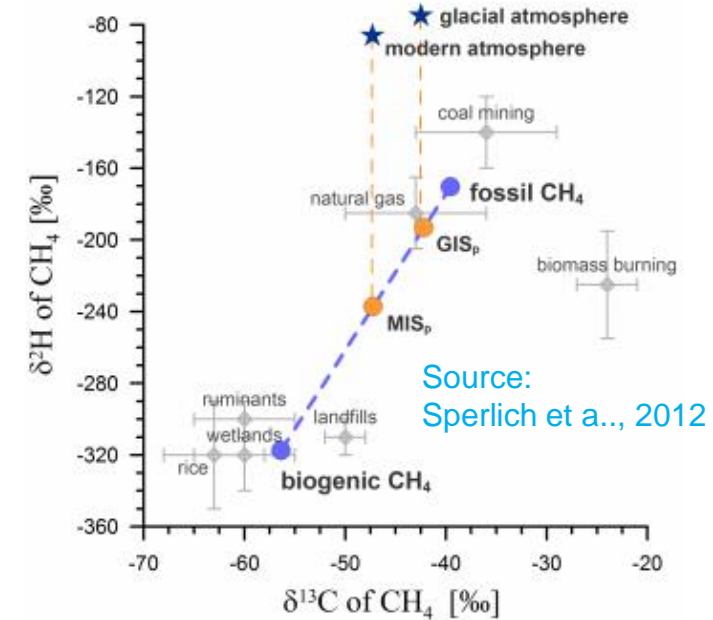
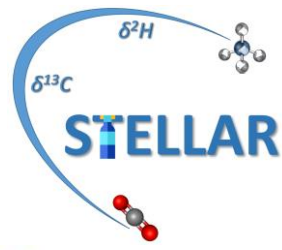
High purity CH<sub>4</sub> (5.5 purity) from fossil origin is available.

## Biogenic

Consulted other NMI's and several companies and organizations.  
 Feedback: CH<sub>4</sub> from Bio-CNG or Bio LNG seem good candidates with respect to low levels of impurities and the required pressure.

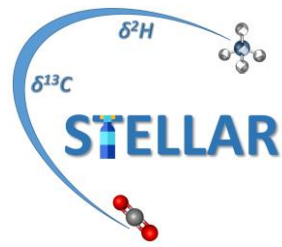
Some producers were contacted. Status:

- 1 x required volume too small
- 1 x facility will become available later this year
- 2 x still under discussion





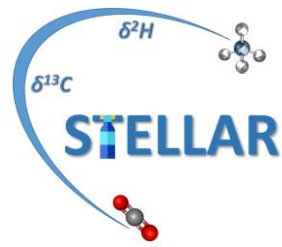
# Activities



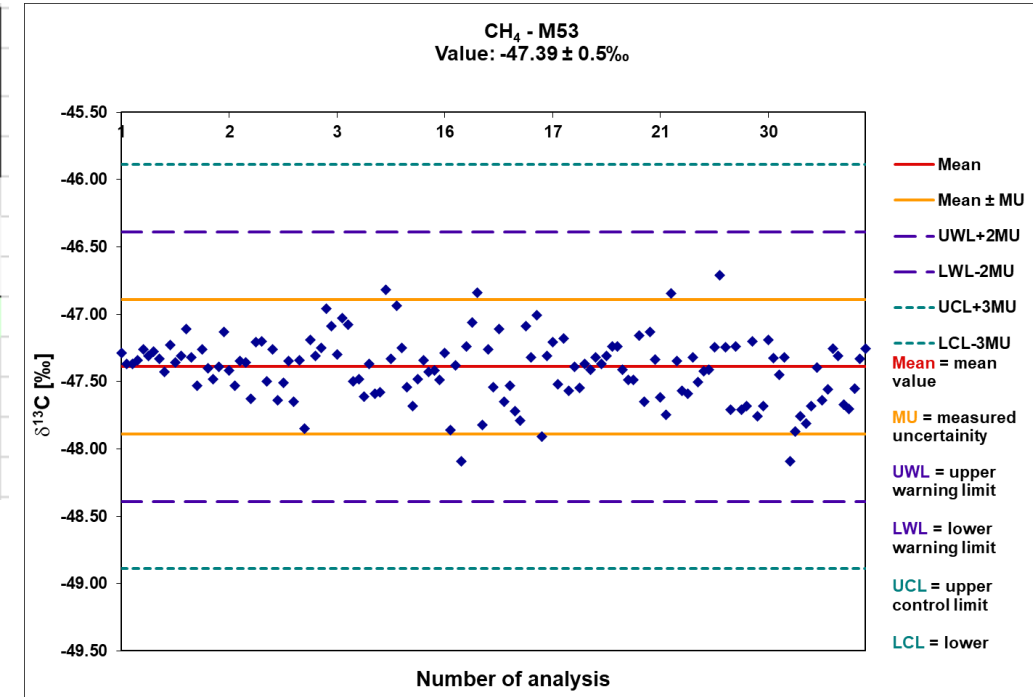
## Available methane sources in Slovenia

Provider	Purity	Cylinder volume [L]	Pressure [bar]	Source	Isotopic composition
Messer	4.5 5.5	50	200	Synthetic	Unknown
Istrabenz (SIAD)	2 2.5 3.5 4.5 5.5 1.7 (fossil gas)	10 20 40 (1 L higher purities only)	200 (1 L: 150)	Natural	$\delta^{13}\text{C} \approx$ from -40 to -60‰ $\delta^2\text{H}$ is unknown
GTG (Linde)	3.5 4.5 5.5	2 (except 3.5) 10 50	200 (2 L: 100)	Unknown	Unknown
TPJ	3.5 From pipeline	50	200	Synthetic	Unknown

## Methane measurements on Europa Scientific 20-20 IRMS with ANCA TG



	Repr.	Bias				
	$u(R_w) = S_{RW}$	Ref value	$U (k=2)$	$u_c$	Lab result	$bias_i$
M53	0.26	-47.39	0.5	0.25	-47.41	-0.02
Če je samo en CRM:						
		$u(C_{ref})$	$RMS_{bias}$	n	s_bias	$u(bias)$
		0.25	0.02	166	0.2554511	0.251569
<b>Result:</b>						
$u_c =$	0.36 ‰					
$U (k=2) =$	0.72 ‰					

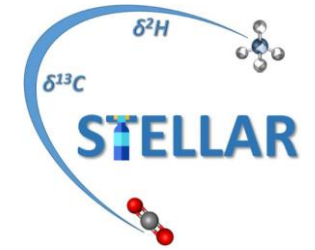


Further activities:

optimization of the  $\delta^{13}C$  determination in CH<sub>4</sub> using EA/IRMS

establishment of the system for  $\delta^2H$  and  $\delta^{13}C$  determination in CH<sub>4</sub> by EA/TC/IRMS

(Sperlich et al., 2016; 2020)



# Air Liquide progress in WP2

## 1.- Pure gases N<sub>2</sub>, O<sub>2</sub> & Ar 50 L 200 bar

- 12 cyl N<sub>2</sub> + 3 cyl O<sub>2</sub> + 1 Cyl Ar sent to NPL Nov 2020
- 14 cyl N<sub>2</sub> waiting for sending to NPL in the Spanish custom
- 8 cyl N<sub>2</sub> + 1 cyl O<sub>2</sub> waiting in Air Liquide for sending to VSL

## 2.- Pure CH<sub>4</sub>

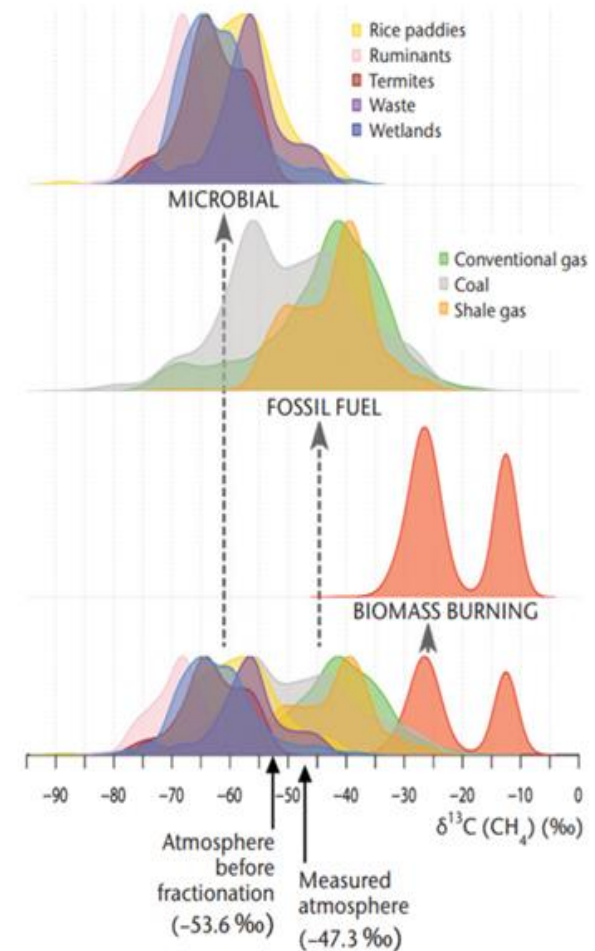
- Two cylinders from same lot 99,999% purity Air Liquide production (fossil source):
  - 1 cyl waiting for sending to NPL in the Spanish custom
  - 1 cyl waiting in Air Liquide for sending to VSL
- One sample of CH<sub>4</sub> sent to Max-Planck (Heiko) for isotope ratio analysis.
- Looking for other possible CH<sub>4</sub> sources. Not found until now: purity and pressure not enough.

## 3.- Empty cylinders with internal treatment

- 26 B10 l cyl prepared and waiting for sending to NPL in the Spanish custom (NPLvalve)
- 20 B05 l and 2 B10 l cyl with DIN-1 or UNI 11144-5 valves for VSL, TUBITAK and INRIM in preparation for Aculife III Megalong treatment
- 2 B10 l cyl with Aculife IV prepared and waiting for sending to VSL.

# Task 2.1/2.2 CH<sub>4</sub>-in-air gas reference materials

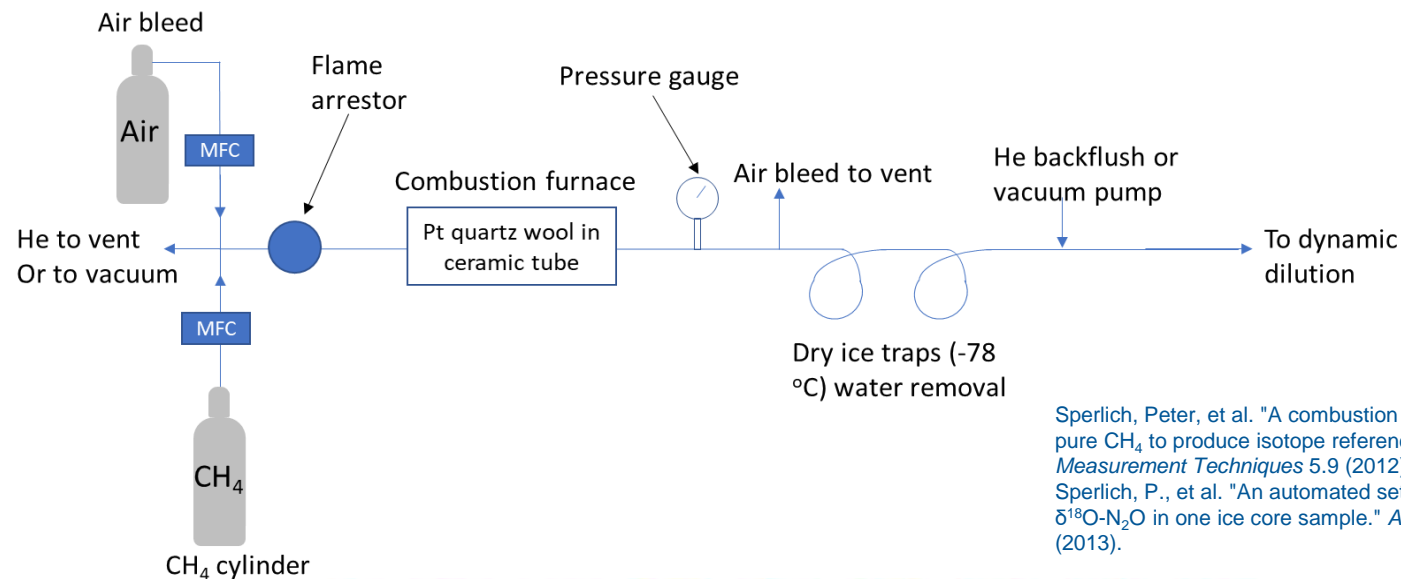
- Obtained one pure CH<sub>4</sub> fossil source, awaiting assignment of  $\delta^{13}\text{C}-\text{CH}_4$  and  $\delta\text{D}-\text{CH}_4$  at Max Planck iso-lab. NPL have a second CH<sub>4</sub> source with a  $\delta^{13}\text{C}-\text{CH}_4$  of near measured ambient  $\delta^{13}\text{C}-\text{CH}_4$ . Still searching for a biomethane source and investigating spiking to achieve full range.
- Developed preparation plan for the dilution of pure CH<sub>4</sub> to 650  $\mu\text{mol mol}^{-1}$  and 1.95  $\mu\text{mol mol}^{-1}$  in synthetic air with the lowest gravimetric uncertainty of 0.082 %  $k=1$  for the ambient amount fractions
- Matrix gases - impurity analysis is underway of Ar, O<sub>2</sub> and N<sub>2</sub> for the Air Liquide matrix gases
- A full uncertainty budget is being prepared for the amount fraction of CH<sub>4</sub> reference materials with 1  $\mu\text{mol mol}^{-1}$  Kr – is dependant on exact matrix gas requirements for WP3



# Task 2.3: Linking to the $\delta^{13}\text{C}\text{-CO}_2$ reference materials for an independent assessment of the accuracy and uncertainty of $\delta^{13}\text{C}\text{-CH}_4$ reference materials

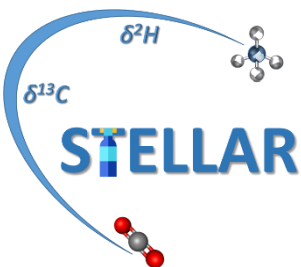
A facility will be developed for the conversion of  $\text{CH}_4 \Rightarrow \text{CO}_2$  and the sensitivities and contribution to uncertainty of this process will be fully characterised. The target uncertainty is 0.1 ‰ for  $\delta^{13}\text{C}\text{-CO}_2$ .

- Developing a  $\text{CH}_4$  to  $\text{CO}_2$  combustion facility based on literature by Sperlich *et al.*
- Design of a dynamic dilution of the  $\text{CO}_2$  output to  $410 \mu\text{mol mol}^{-1}$  using a two stage system based on mass flow controllers and mol blocs for direct measurement at  $410 \mu\text{mol mol}^{-1}$  by CRDS.



Sperlich, Peter, et al. "A combustion setup to precisely reference  $\delta^{13}\text{C}$  and  $\delta^2\text{H}$  isotope ratios of pure  $\text{CH}_4$  to produce isotope reference gases of  $\delta^{13}\text{C}\text{-CH}_4$  in synthetic air." *Atmospheric Measurement Techniques* 5.9 (2012): 2227-2236.

Sperlich, P., et al. "An automated setup to measure paleoatmospheric  $\delta^{13}\text{C}\text{-CH}_4$ ,  $\delta^{15}\text{N}\text{-N}_2\text{O}$  and  $\delta^{18}\text{O}\text{-N}_2\text{O}$  in one ice core sample." *Atmospheric Measurement Techniques Discussions* 6.1 (2013).



**Thank you for your attention!**

**Any questions or remarks?**

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