



Hyperspectral techniques for air pollutant detection and quantification

16ENV08 IMPRESS 2: Metrology for Air Pollutants Emissions Presented by: Guillermo Guarnizo (UC3M)





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Stakeholder workshop via MS Teams

Outline

- Introduction to hyperspectral techniques
- Survey of the latest technologies related to hyperspectral detection of pollutant emissions
- Development of calibration methods and set-ups: Intercomparison of calibration methods
- Preparation of reference gas mixtures for validation activities
- Validation of hyperspectral methods for identification and quantification
- Conclusions





Hyperspectral techniques: Fundamentals

- Kirchhoff law (1860): Emissivity equals absorptance $\rightarrow \alpha(\lambda) = \varepsilon(\lambda)$
- Absorption in gases: It can be measured by the radiance through the pollutant → ε = α = 1 − τ



• Lambert-Beer law: $\tau(\lambda) = e^{-a(\lambda)CL}$

where

- a = Absorptivity (gas feature)
- C = Concentration
- L = Optical path
- CL = Column density (ppm.m)

• Dispersive or interferometer systems IFTS: FTIR based on Michelson interferometer + MIR InSb Camera



• Detection of continuous spectra



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Survey of the latest technologies related to hyperspectral detection of pollutant emissions

- Most pollutant gases are very selective absorbers and emitters of infrared (IR) radiation
- Their concentration in air can be quantified → High spectral resolution IR data (e.g. HITRAN or PNNL IR database)
- Overview of recent literature:
 - Books focused on principles of hyperspectral detection
 - Review and peer-reviewed papers
- Research groups working on hyperspectral techniques
- Available commercial systems







Calibration methods for hyperspectral systems **Procedure for passive systems (UC3M)** Radiometric model



Measurement: $\mathcal{L}^{in} = \mathcal{L}_m = \mathcal{L}^{\mathcal{B}}(T_b) \cdot \varepsilon_b \cdot \tau_{a_1} \tau_g \tau_{a_2} + \mathcal{L}^{\mathcal{B}}(T_g) \cdot (1 - \tau_g) \tau_{a_2}$ Reference: $\mathcal{L}^{in} = \mathcal{L}_r = \mathcal{L}^{\mathcal{B}}(T_b) \cdot \varepsilon_b \cdot \tau_{a_1} \tau_{g_0} \tau_{a_2}$ $\tau_{nom} \equiv \frac{\mathcal{L}_m}{\mathcal{L}_r} = \tau_g + \frac{\mathcal{L}^{\mathcal{B}}(T_g)}{\mathcal{L}^{\mathcal{B}}(T_b)} \cdot (1 - \tau_g) \cdot \frac{1}{\varepsilon_b \tau_g} \equiv \tau_g + \tau'$

 $\Rightarrow \tau(\lambda) = e^{-a(\lambda)CL} \rightarrow \text{Lambert-Beer law provides CL} = \text{column density}$

Uncertainty estimation in gas column density due to:

- Measurement noise
- In absorption mode: Error due to gas emission Better for cold air pollutants
- In emission mode: Error due to radiometric calibration of IFTS →



CEM Calibration facilities

EURAM

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Better for hot air pollutants

Preparation of reference gas mixtures

Mixtures preparation

- ✓ Following CEM gravimetric method
- ✓ ISO 6142-1 Preparation of calibration gas mixtures
- ✓ Analytical check ISO 6143, to ensure prepared concentrations

Compounds	UC ₃ M	NPL & VSL
CH ₄	(602,4 ± 1,4) µmol/mol	(200,0 \pm 1,5) µmol/mol
C_3H_8	(499,9 ± 1,2) μmol/mol	$(200,0 \pm 1,0) \mu mol/mol$
N ₂ O	(250,4 ± 2,5) µmol/mol	$(200,5\pm2,0) \mu\text{mol/mol}$









Matrix: synthetic air $(20\% O_2, 80\% N_2)$



Preparation of reference gas mixtures



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Identification and quantification of air pollutants: UC3M



Air pollutant stand-off measurements

- $\circ~$ MIR region: 1,5 μm to 5,5 μm
- Spectral resolution: 1 cm⁻¹
- 2 set of acquired data: air pollutant and reference gas
- 3 gases studied: Methane (CH_4), Nitrous Oxide (N_2O) and Propane (C_3H_8)





Features of hyperspectral method:

- ✓ A spectrum for each pixel of the image
- ✓ Transversal uniformity in gas cell verified
- Robust fitting algorithm to retrieve physical variables: Column density and temperature
- Possibility to establish a minimal concentration value for quantification





Identification and quantification of air pollutants: UC3M





Column density maps (ppm.m) for the three air pollutants agreed by the partners



Pollutant Gas	Concentration	Column density	Retrieved column
	(ppm)	(ppm.m)*	density (ppm.m)**
Methane (CH ₄)	600	258	250 ± 9,8
Nitrous oxide (N ₂ O)	250	107,5	112,3 ± 1,4
Propane (C ₃ H ₈)	500	215	199,7 ± 3,8

* Calculated values based on the gas cell of 43 cm used on measurements. ** Retrieved values from a 7 x 7 pixels central area of the hyperspectral image.



Identification and quantification of air pollutants: VSL

- VSL designed compact OPO (tuneable IR source)
- Parts have been ordered and system assembled. Unfortunately system not working properly (Nd-YVO4 laser unstable)









Identification and quantification of air pollutants: VSL

• As designed OPO did not work, a different route was chosen using a Nd-YAG pumped OPO



Features OPO system

- Wavelength range 2.3-5.1 µm
- Output power up to 2.5 Watt
- Continuous wave operation
- Rapid tuning via etalon (up to 10's of Hz) or pump laser (up to 1 kHz)



Identification and quantification of air pollutants: VSL

• OPO can enhance the thermal contrast in measurements using hyperspectral imagers and so improve quantification of gas concentrations



Figure 2 Direct absorption measurement of the HCl line centred at 2963.29 cm⁻¹. The OPO was tuned at a rate of 20 Hz over the absorption line.





Figure 3 Tuning of the OPO using only etalon tuning.



Conclusions

- A rigorous survey on novel hyperspectral techniques has been drafted. Books, papers and commercial systems are included in the final document
- Traceable calibration methods for hyperspectral imagers have been established for passive systems
- A defined group of air pollutants for validation measurements has been prepared and certified by CEM
- Excellent results in quantification of pollutants have been achieved by UC3M with an improved hyperspectral method
- The OPO-based method proposed by VSL are good enough in spite of the problems they had with Nd-YVO4 laser unstable









Thanks for your attention

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